Preparation of Benzocycloheptene Derivatives from Zirconacyclopentadienes

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ABSTRACT





Benzocycloheptene derivatives are attractive in theoretical chemistry,¹ pharmaceutical sciences,² and coordination chemistry.³ They have been prepared by several methods such as the enlargement of six-membered rings⁴ and cyclization reactions.⁵ However, preparation of benzocycloheptenes having substituted seven-membered rings has been very limited. In this paper we would like to report a novel type

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10.1021/ol005533r CCC: \$19.00 © 2000 American Chemical Society Published on Web 04/06/2000 of [4 + 3] coupling reaction of zirconacyclopentadienes with 2-iodobenzyl or 2-iodobenzoyl halides.

Zirconacyclopentadienes are versatile intermediates that can be easily prepared symmetrically or unsymmetrically from alkynes or diynes. Recently we have reported the [4 + 4], [4 + 5], and [4 + 2] coupling reaction of zirconacyclopentadienes with bis(bromomethyl)arenes or a dihalobenzene derivative, giving eight-, nine-, or six-membered ring compounds.^{6,7} During the course of our study, we found that 2-iodobenzyl or 2-iodobenzoyl halides can be a useful threecarbon unit for the [4 + 3] coupling of zirconacyclopentadienes as shown in Scheme 1.



Typically, zirconacyclopentadienes were prepared by coupling of two alkynes with Cp_2ZrR_2 (R = Bu, Et, or Ph)

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Table 1. Reaction of Zirconacyclopentadienes with 2-Iodobenzyl Halide



according to the literature.⁸ To a solution of tetraethylzirconacyclopentadiene **1a** (1 mmol) prepared from Cp_2ZrBu_2 (1.2 mmol) and 3-hexyne (2 mmol) in THF was added 2-iodobenzyl halide **2a** (1 mmol). The reaction was carried out at 50 °C in the presence of a stoichiometric amount of CuCl (2 mmol) and dimethylpropyleneurea (DMPU, 2.5 mmol). After 1 h, the desired product, 1,2,3,4-tetraethylbenzoheptene, **3a**, was obtained in 70% yield after purification by column chromatography. Results are summarized in Table 1.

It was found that zirconacyclopentadienes 1a-d, either monocyclic or bicyclic, underwent the reaction well and gave the corresponding products in high yields. Heterobicyclic zirconacyclopentadiene $1f^9$ gave the product 3f. However, the reaction did not proceed for the zirconacyclopentadienes with aryl or silyl substituents in the α -position.

It is noteworthy that in the case of a zirconaindene **1e**, which has a benzene moiety in the zirconacycle, the reaction

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4e

proceeded, and it is in contrast to the case with phenylsubstituted zirconacyclopentadienes. Reaction of the unsymmetrical zirconaindene with 2a might afford a mixture of two isomers, 3e and 4e. As a matter of fact, only 3e was obtained as shown in Scheme 2. A plausible mechanism of the reaction of 1e with 2a is shown in Scheme 3. We have reported the transmetalation of zirconacyclopentadienes or zirconaindenes to copper.¹⁰ An alkyl-substituted sp² carbon attached to zirconium reacts with CuCl faster than the aryl carbon attached to zirconium. The more reactive carbon attached to copper reacts with the benzyl carbon of 2a first. Then, intramolecular coupling of the arylzirconium moiety with the aryl iodide moiety of 2a proceeds in the presence of CuCl. This mechanism rationalizes the selective formation of 3e. As shown in Table 1, 2-iodo-3,4,5,6-tetramethylbenzyliodide can be used for the formation of benzocycloheptenes, although the isolated yield was only 23%.

Similar [4 + 3] coupling reaction of zirconacyclopentadiene was observed in the reaction with 2-iodobenzoyl





chloride **5**. Under the same conditions the benzocycloheptenone derivatives **6a,b** were obtained as shown in Scheme 4.



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Supporting Information Available: Characterization and ¹H and ¹³C NMR spectra for **3a**–**g** and **6a,b**. This material is available free of charge via the Internet at http://pubs.acs.org. OL005533R

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